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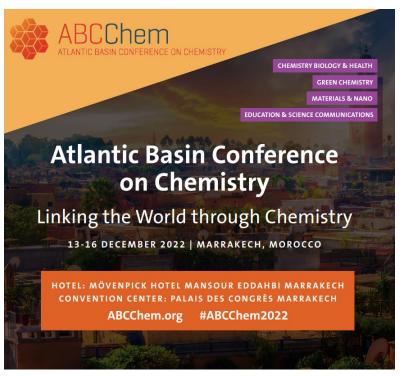
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1 Careers in Academia
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ACS Scholar Adunoluwa Obisesan

BS, Massachusetts Institute of Technology, June 2021 (Chemical-biological Engineering, Computer Science & Molecular Biology)

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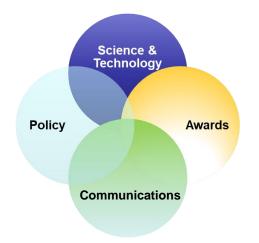
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Wed., Aug. 17, 2022 | 2:00pm-3:00pm ET

Catalizadores de la Reducción de CO₂ con Gran Promesa

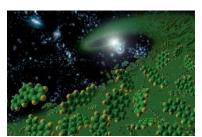
Co-produced with the Sociedad Química de México



Thurs., Aug. 18, 2022 | 2:00pm-3:00pm ET

An Evolutionary Mystery: Mirror Asymmetry in Life and in Space

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Wed., Aug. 31, 2022 | 2:00pm-3:15pm ET

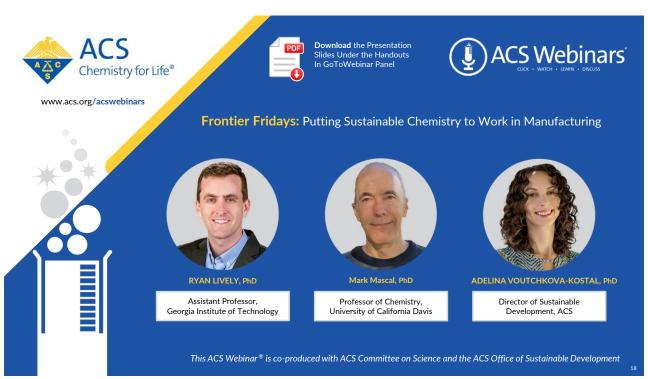
How We Study Molecules in Space

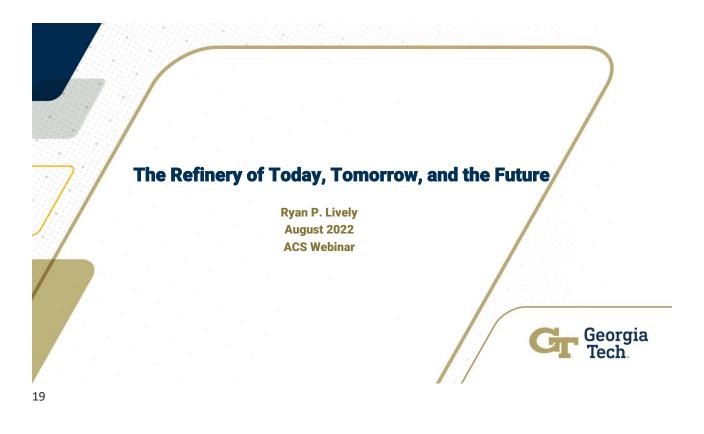
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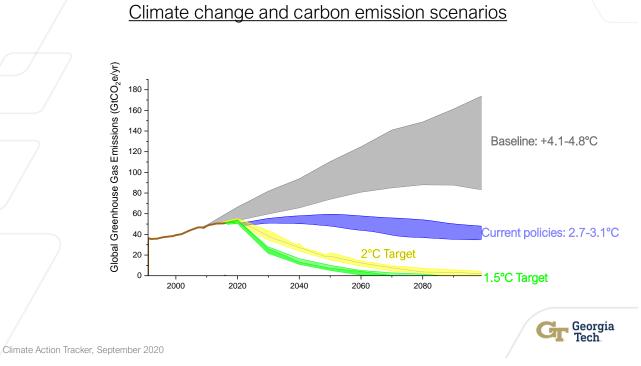
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Audience Survey Question

ANSWER THE QUESTION ON THE INTERACTIVE SCREEN IN ONE MOMENT

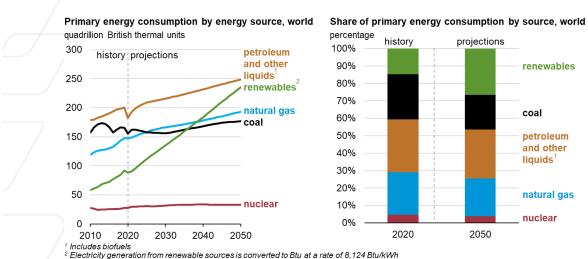
Will we the use of hydrocarbons be by 2050?

- We will use less hydrocarbons than we do today
- We will use about the same amount
- We will use more than we do today

2

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Speculation: population growth in developing regions will more than offset growth in renewables



The future may require *more* hydrocarbons, not less! ...but with significantly reduced carbon emissions...

U.S. Energy Information Administration, International Energy Outlook 2021

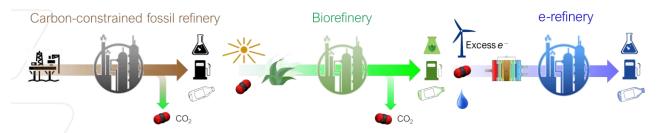
Georgia Tech

^{*} If your answer differs greatly from the choices above tell us in the chat!

The Refinery of Today, Tomorrow, and the Future:

Producing *more* fuels, chemicals, and plastics with dramatically *lower* CO₂ emissions

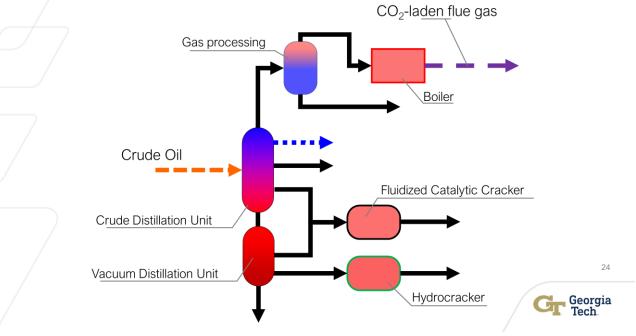
The three low-carbon archetypes:



Speculation: existing capital equipment, permitting, zoning, and regulations provides strong driving force to evolve existing systems and switch the fuel source



Speculation: refinery flowsheet will reconfigure during energy transition



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Speculation: advanced separations will play a critical role in future refineries

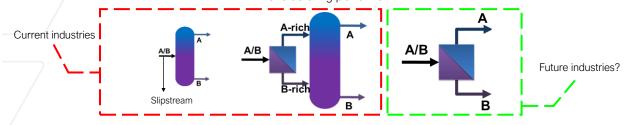
+ CO₂ removal from air

RP Lively, AIChE J 2021

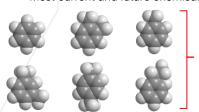
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Membranes play a key role in the energy transition

Modular membrane systems can reduce carbon/energy intensity of existing manufacturing paradigms and enable futuristic manufacturing platforms!

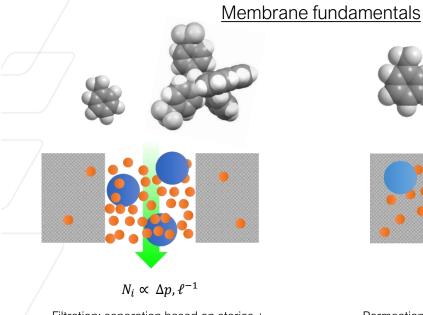


Most current and future chemicals, fuels, and commodities involve the separation of "small" molecules < 300 Da



Real mixtures are often concentrated liquids and are complex

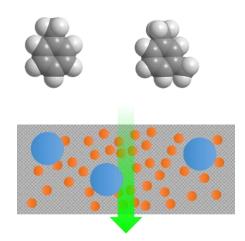




Filtration; separation based on sterics + interaction with membrane surface

P. Marchetti, Chem. Rev. 2014, 114(21), 10735-10806

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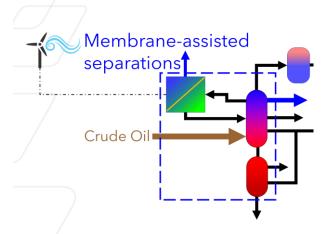
 $N_i \propto \Delta C_i^m, \ell^{-1}$

Permeation; $\mathbb{P}_i = D_i \cdot \mathbb{S}_i$, separation based on sorption into membrane bulk and guest diffusivity

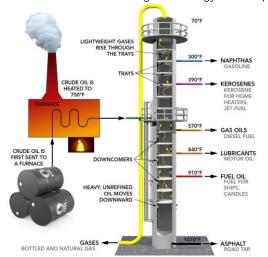
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Examples of emerging separations technologies: crude oil fractionation

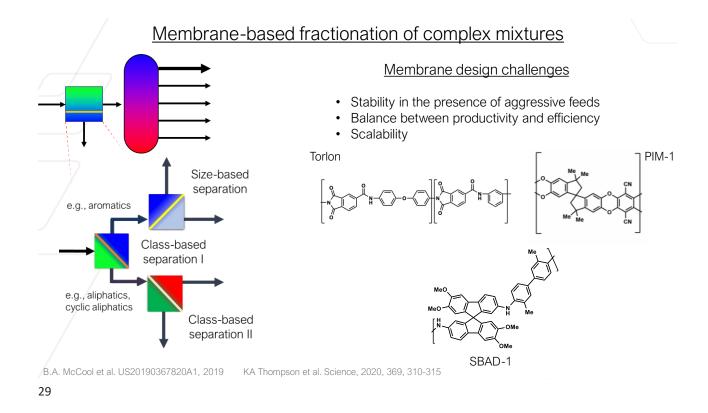
Crude oil fractionation: ~0.5-1% global energy consumption



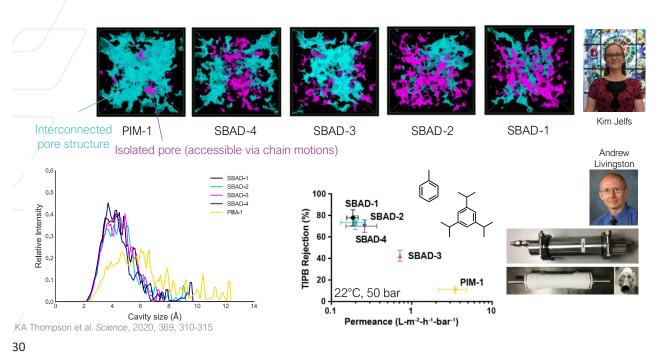
Energy and carbon intensity reductions by 1.5-2x!



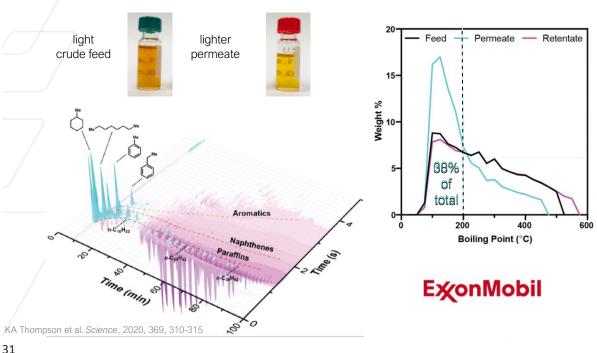




Relating polymer structure to membrane microstructure to guest transport



<u>Complex mixture separations in action – real mixtures</u>



Review and Final Thoughts

- Existing refineries may "retool" to incorporate electrically-driven processes, bio-based feedstocks, and advanced separations
- · Some "e-refinery" pilots are being constructed





- Membranes have the potential to enable more efficient refining processes (chemical, petro, bio, e-, ...)
- Identification of key win-win separations within the various refining processes (>100) remains a challenge
- How do we rapidly transition these new concepts out of the lab and into the field?





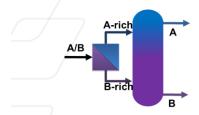


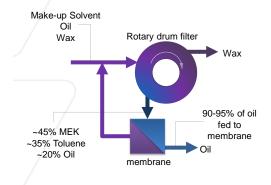




Many opportunities and challenges await in separations science & engineering!

Key observations from successful membrane demonstrations





RM Gould et al., Environmental Prog. 2004, 20(1), 12-16



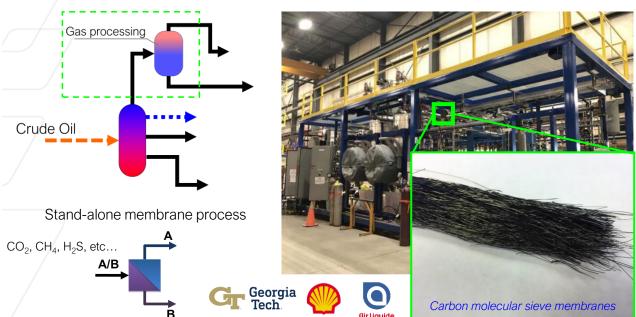
- · 20% reduction in process energy intensity
- 20,000 tons/year reduction in GHG emissions
- Reduced VOC emissions by 125 tons/year
- Reduced water usage by 4 million gal/day
- Payback period <1 year; >600 days continuous operation







Pilot applications for membranes in chemical and petrochemical plants



Koros et al. Ind. Eng. Chem. Res. 2017, 56(37), 10482-10490

Lima Touma et al., Offshore Tech. Conf. 2019, OTC-29913-MS

The Story of 5-(Chloromethyl)furfural (CMF)



Mark Mascal UC Davis

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What is Chemistry's response to the climate crisis?



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Context: Green Chemistry, then and now

- 1. Prevent waste
- 2. Atom Economy
- 3. Less Hazardous Synthesis
- 4. Design Benign Chemicals
- 5. Benign Solvents & Auxiliaries
- 6. Design for Energy Efficiency
- 7. Use of Renewable Feedstocks
- 8. Reduce Derivatives
- 9. Catalysis (vs. Stoichiometric)
- 10. Design for Degradation
- 11. Real-Time Analysis for Pollution Prevention
- 12. Inherently Benign Chemistry for Accident Prevention



Fewer hazards, less waste

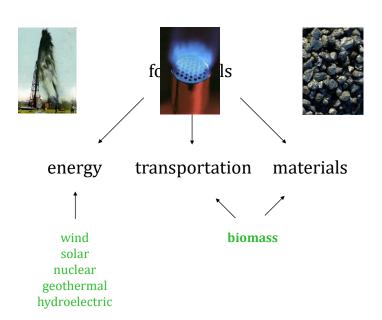
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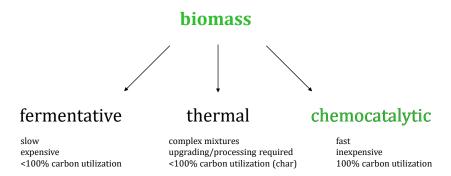
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Biomass processing



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2006 UC Davis - Chevron

Joint Research Agreement

Biofuels Research: \$25 million/5 years

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My Chevron Project

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H. J. H. Fenton, M. Gostling, J. Chem. Soc. Trans. 1899, 75, 423

XLI.—Bromomethylfurfuraldehyde.

By Henry J. Horstman Fenton, M.A., and Miss Mildred Gostling, B.Sc., Bathurst Student of Newnham College.

In a previous communication (Trans., 1898, 73, 554), it was shown by the present authors that certain carbohydrates, when treated with hydrogen bromide in ethereal solution, give an intense and beautiful purple colour. It was suggested that the reaction would probably be of service in distinguishing certain typical classes of carbohydrates, and that it might be employed for the detection of these in natural products.

The examination of a considerable number of carbohydrates of various classes led to the following generalisations.

- 1.—With ketohexoses (lavulose, sorbose), the purple colour appears after a few minutes, and becomes very intense in about one hour; and substances which, by hydrolysis, are capable of giving rise to these compounds (cane sugar, inulin) give a similar effect, only rather more slowly, the maximum colour being attained in about two hours.
- 2.—Hexaldoses (dextrose, galactose), and substances capable of giving rise to them (milk sugar, maltose, dextrin), show, in the first instance, various shades of yellow, brown, or red, and it is only after long standing that a purple colour is apparent. This colour, however, never approaches in brilliancy or intensity that obtained with substances mentioned in 1.

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5-(Chloromethyl)furfural (CMF)

- CMF is a stable, low-melting solid (mp 37-38 °C)
- functional equivalent of and interconvertible with HMF
- product isolation a simple matter of evaporating solvent
- flow systems reduce reaction time to minutes

M. Mascal, E.B. Nikitin, Angew. Chem. Int. Ed. 2008, 47, 7924

M. Mascal, E.B Nikitin, ChemSusChem 2009, 2, 859

M. Mascal, US Patent 2010, 7,829,732

Review: M. Mascal, ACS Sustainable Chem. Eng. 2019, 7, 5588

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CMF reactivity

C-electrophile
C-nucleophile

H. Miao, N. Shevchenko, A. L. Otsuki, M. Mascal, ChemSusChem 2021, 14, 303

H. Miao, H. Ling, N. Shevchenko, M. Mascal, Organometallics 2021, 40, in press

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CMF as a platform molecule



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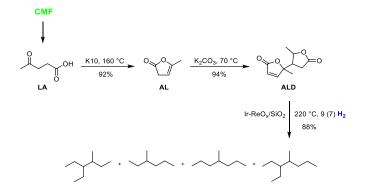
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The CMF biofuel family tree

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Fuels: Cellulosic gasoline 1



M. Mascal, S. Dutta, I. Gandarias, Angew. Chem. Int. Ed. 2014, 53, 1885

F. Chang, S. Dutta, M. Mascal, ChemCatChem 2017, 9, 2622

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Fuels: Cellulosic gasoline 2

L. Wu, M. Mascal, T. J. Farmer, S. Perocheau Arnaud, M.-A. Wong Chang, ChemSusChem 2017, 10, 166

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Fuels: Cellulosic gasoline 3

Z. Li, A. L. Otsuki, M. Mascal, Green Chemistry, 2018, 20, 3804

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Fuels: Aromatics and oxygenates

S. Dutta, M. Mascal, ChemSusChem 2014, 7, 3028

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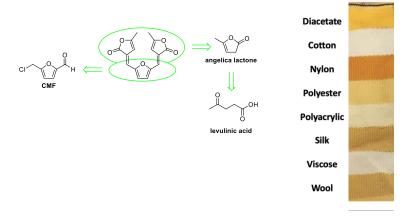
The CMF chemical family tree

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Biobased synthetic dyes from CMF



J. Saska, Z. Li, A. L. Otsuki, J. Wei, J. C. Fettinger, M. Mascal Angew. Chem. Int. Ed. 2019, 58, 17293

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The CMF monomer family tree

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The Meeting



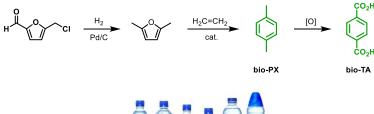


September 6, 2010

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Monomers: terephthalic acid















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Monomers: terephthalic acid





Origin Materials (ORGN) market cap >\$1B

Origin 1

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Monomers: terephthalic acid







Origin 2 – Geismar, LA Construction start 2023, commissioning mid-2025

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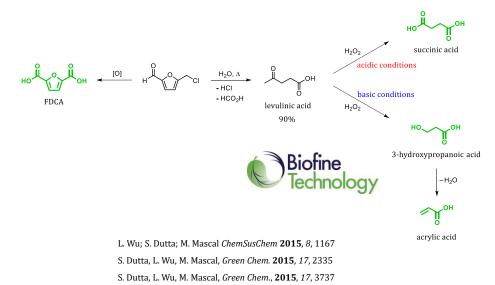
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EPA Green Chemistry Challenge Award



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Monomers: succinic, 3-HPA, and bio-acrylic acid



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Other CMF commercialization

Mercurius - Australia - cellulosic jet fuel

 \mathbf{xF} - Albuquerque, New Mexico - 408^{TM} fuel additives

Furanica - Davis, California - furan fatty acids









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Acknowledgements – coworkers

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Dr Saikat Dutta Maria Angelica Wong Chang

Dr David Lane Zheng Li

Dr Linglin Wu Edwin Naranjo-Valles

Dr Inaki Gandarias Andrew Otsuki

















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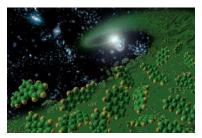
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